



Fig. 2. Diagram illustrating the  $P, T$  fields of different mineral assemblages in pyrolite III composition. The figures 1%  $\text{Al}_2\text{O}_3$ , 2%  $\text{Al}_2\text{O}_3$  etc. refer to the  $\text{Al}_2\text{O}_3$  content of orthopyroxene in equilibrium with garnet in the garnet pyrolite field. The oceanic and Precambrian shield geotherms are those illustrated by Ringwood et al. [13].

simple 2-component systems. This effect is attributed to differences in garnet and pyroxene composition, possibly relating particularly to Ca and Cr contents and the presence of clinopyroxene, between the two component and complex systems. The persistence of spinel to around 1300°C in the pyrolite composition contrasts with its anticipated absence above 1000°C predicted by Ringwood et al. [13] and is also attributed to chemical or coupled reaction effects in the complex composition.

#### 4. MINERALOGY IN A PYROLITE UPPER MANTLE.

In the following discussion, it is assumed that the average composition of the upper mantle beneath

oceanic and geologically "young" regions is that of pyrolite. In continental regions and particularly in Precambrian shield regions, there is probably a much greater proportion of refractory residual peridotite at shallower levels and the upper mantle may be chemically zoned [3,4]. Mineralogical variation in the upper mantle is determined by the intersection of geothermal gradients with the stability fields of fig. 1. The variation of temperature in the continental and oceanic crusts and upper mantle has been discussed by Clark [15] and Clark and Ringwood [16]. It is clear from these papers that there is a large difference between oceanic and stable shield geotherms and that this difference takes the form illustrated in fig. 2. Nevertheless, the specific form of the curves, particularly their "convexity" and their closeness of approach to the pyrolite solidus, depend sensitively on knowledge of

